## Class Nine Academic Program 2025

## [New Batch- Online]

## Class & Exam Routine Part- 01 [English Version]

	Live Class- 01	Live Class- 02	Live Exam	
Date & Day	(7:00pm- 8:00pm)	(9:30pm-10:30pm)	Online: 9:00am to 11:55pm	
20 April 2025 (Sunday)	(B-03); Biology: Chapter-02	(HM-09); H.Math: Chapter-02	Basic Introductory Exam	
21 April 2025 (Monday)	(B-04); Biology: Chapter-02	(M-09); Math: Chapter-03	Daily Live Exam <b>(B-03) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-09) MCQ</b> (10×1=10); 10 min.	
22 April 2025 (Tuesday)	(C-03); Chemistry: Chapter-02	( <mark>HM-10); H.</mark> Math: Chapter-02	Daily Live Exam <b>(B-04) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-09) MCQ</b> (10×1=10); 10 min.	
23 April 2025 (Wednesday)	(C-04); Chemistry: Chapter-02	(P-03); Physics: Chapter-02	Daily Live Exam <b>(C-03) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-10) MCQ</b> (10×1=10); 10 min.	
24 April 2025 (Thursday)	(M-10); Math: Chapter-03	(P-04); Physics: Chapter-02	Daily Live Exam <b>(C-04) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-03) MCQ</b> (10×1=10); 10 min.	
27 April 2025 (Sunday)	(B-05); Biology: Chapter-02	(HM-11); H.Math: Chapter-02	Daily Live Exam <b>(M-10) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-04) MCQ</b> (10×1=10); 10 min.	
28 April 2025 (Monday)	(B-06); Biology: Chapter-02	(M-11); Math: Chapter-03	Daily Live Exam <b>(B-05) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-11) MCQ</b> (10×1=10); 10 min.	
29 April 2025 (Tuesday)	(C-05); Chemistry: Chapter-02	(HM-12); H.Math: Chapter-02	Daily Live Exam <b>(B-06) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-11) MCQ</b> (10×1=10); 10 min.	
30 April 2025 (Wednesday)	(C-06); Chemistry: Chapter-02	(P-05); Physics: Chapter-02	Daily Live Exam <b>(C-05) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-12) MCQ</b> (10×1=10); 10 min.	
01 May 2025 (Thursday)	(M-12); Math: Chapt <mark>er-03</mark>	(P-06); Physics: Chapter-02	Daily Live Exam <b>(C-06) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-05) MCQ</b> (10×1=10); 10 min.	
02 May 2025 (Friday) (	Chapter Wise Exam: <mark>Chemistry-</mark> Chapte	<mark>er- 02 (States o</mark> f Matter) MCQ (10×1=10	0); 10 min & CQ/ Written (30 marks); 1 hour.	
04 May 2025 (Sunday)	(B-07); Biology: Cha <mark>pter-02</mark>	(HM-13); H.Math: Chapter-02	Daily Live Exam <b>(M-12) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-06) MCQ</b> (10×1=10); 10 min.	
05 May 2025 (Monday)	(B-08); Biology: Chap <mark>ter-02</mark>	(M-13); Math: Chapter-03	Daily Live Exam <b>(B-07) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-13) MCQ</b> (10×1=10); 10 min.	
06 May 2025 (Tuesday)	(C-07); Chemistry: Chapter-03	(HM-14); H.Math: Chapter-02	Daily Live Exam <b>(B-08) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-13) MCQ</b> (10×1=10); 10 min.	
07 May 2025 (Wednesday)	(C-08); Chemistry: Chapter-03	(P-07); Physics: Chapter-02	Daily Live Exam <b>(C-07) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-14) MCQ</b> (10×1=10); 10 min.	
08 May 2025 (Thursday)	(M-14); Math: Chapter-03	(P-08); Physics: Chapter-02	Daily Live Exam <b>(C-08) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-07) MCQ</b> (10×1=10); 10 min.	
09 May 2025 (Friday) <mark>C</mark>	hapter Wise Exam: H.Math- Chapter- O	2 (Algebraic Expression) MCQ (10×1=1	0); 10 min & CQ/ Written (30 marks); 1 hour.	
11 May 2025 (Sunday)	(B-09); Biology: Chapter-02	(HM-15); H.Math: Chapter-03	Daily Live Exam <b>(M-14) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-08) MCQ</b> (10×1=10); 10 min.	
12 May 2025 (Monday)	(B-10); Biology: Chapter-02	(M-15); Math: Chapter-03	Daily Live Exam <b>(B-09) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-15) MCQ</b> (10×1=10); 10 min.	
13 May 2025 (Tuesday)	(C-09); Chemistry: Chapter-03	(HM-16); H.Math: Chapter-03	Daily Live Exam <b>(B-10) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-15) MCQ</b> (10×1=10); 10 min.	
14 May 2025 (Wednesday)	(C-10); Chemistry: Chapter-03	(P-09); Physics: Chapter-02	Daily Live Exam <b>(C-09) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-16) MCQ</b> (10×1=10); 10 min.	
15 May 2025 (Thursday)	(M-16); Math: Chapter-03	(P-10); Physics: Chapter-02	Daily Live Exam <b>(C-10) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-09) MCQ</b> (10×1=10); 10 min.	
16 May 2025 (Friday) Chapte	er Wise Exam: Biology- Chapter- 02 (Ce	lls and Tissues of Organisms) MCQ (1	0×1=10); 10 min & CQ/ Written (30 marks); 1 hour.	
18 May 2025 (Sunday)	(B-01); Biology: Chapter-01	(HM-17); H.Math: Chapter-03	Daily Live Exam <b>(M-16) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-10) MCQ</b> (10×1=10); 10 min.	
19 May 2025 (Monday)	(B-02); Biology: Chapter-01	(M-01); Math: Chapter-01	Daily Live Exam <b>(B-01) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-17) MCQ</b> (10×1=10); 10 min.	
20 May 2025 (Tuesday)	(C-11); Chemistry: Chapter-03	(HM-18); H.Math: Chapter-03	Daily Live Exam <b>(B-02) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-01) MCQ</b> (10×1=10); 10 min.	
21 May 2025 (Wednesday)	(C-12); Chemistry: Chapter-03	(P-01); Physics: Chapter-01	Daily Live Exam <b>(C-11) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-18) MCQ</b> (10×1=10); 10 min.	
22 May 2025 (Thursday)	(M-02); Math: Chapter-01	(P-02); Physics: Chapter-01	Daily Live Exam <b>(C-12) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-01) MCQ</b> (10×1=10); 10 min.	
23 May 2025 (Fr	iday) Chapter Wise Exam: Physics- Cha	pter- 02 (Motion) MCQ (10×1=10); 10 n	nin & CQ/ Written (30 marks); 1 hour.	
24 May 2025 (Saturday	) Chapter Wise Exam: Math- Chapter- O	3 (Algebraic Expression) MCQ (10×1=	10); 10 min & CQ/ Written (30 marks); 1 hour.	
25 May 2025 (Sunday)	(B-11); Biology: Chapter-03	(HM-19); H.Math: Chapter-03	Daily Live Exam <b>(M-02) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-02) MCQ</b> (10×1=10); 10 min.	

26 May 2025 (Monday)	(B-12); Biology: Chapter-03	(M-03); Math: Chapter-01	Daily Live Exam <b>(B-11) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(HM-19) MCQ</b> (10×1=10); 10 min.	
27 May 2025 (Tuesday)			Daily Live Exam <b>(B-12) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(M-03) MCQ</b> (10×1=10); 10 min.	
28 May 2025 (Wednesday)	(C-01); Chemistry: Chapter-01	(P-11); Physics: Chapter-03		
29 May 2025 (Thursday)	(M-04); Math: Chapter-01	(P-12); Physics: Chapter-03	Daily Live Exam <b>(C-01) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-11) MCQ</b> (10×1=10); 10 min.	
30 May 2025 (Friday)	(C-02); Chemistry: Chapter-01	(HM-20); H.Math: Chapter-03	Daily Live Exam <b>(M-04) MCQ</b> (10×1=10); 10 min. Daily Live Exam <b>(P-12) MCQ</b> (10×1=10); 10 min.	
	Chapter Wise Exam: H.Math- Chapter- 03 (Geometry) MCQ (10×1=10); 10 min & CQ/ Written (30 marks); 1 hour.			
31 May 2025 (Saturday) Chapter Wise Exam: Biology- Chapter- 01 (Lessons of Life) MCQ (10×1=10); 10 min & CQ/ Written (30 marks); 1 hour.				
Next Class & Exam Routine will be published at (Part-02)				

## Online Class & Exam System:

- Scan the QR code below to attend classes and exams or visit online.udvash-unmesh.com and login using the registration number provided.
- You can appear once between 9 am to 11:55pm as per date mentioned in daily exam routine.
- However, for more practice, students can participate in the **Practice Exam** of the same syllabus multiple times.
- Use the Past Class option to view recorded videos and PDFs of daily classes.
- **Q&A** option can be used 24/7 to solve any subject related problems after the class.
- All those admitted in the 'Combo Batch' can participate in the Chapter wise exams online as well as in any nearby branch.
- Join our Facebook group (https://www.facebook.com/groups/class6789.udvashunmesh) to get all information in time.

	Physics		
<u>Chapter</u>	<u>Lecture</u>	<u>Syllabus</u>	
	P-03	Rest and Motion, Different Types of Motion (Linear Motion, Circular Motion, Translational Motion, Periodi <mark>c Motion, Simple Harmonic Motio</mark> n)	
	P-04	Scalars and Vectors Quantities, Distance and Displacement	
Chapter-02	P-05	Speed and Velocity, Acceleration and Deceleration or Retardation	
Motion	P-06	Equations of Motion	
Motion	P-07	Laws of Falling Bodies	
	P-08	Graph related problem	
	P-09	Mathematical problems	
	P-10	Mathematical problems	
<u>Chapter-01</u>	P-01	Physics, Scope of Physics, Development of Physics, Initial Stage, Rising Stage of Science, Introduction to Modern Physics, Contemporary Physics, Contributions of Jagadish Chandra Bose, Objectives of Physics, Unfold the Mystery of Nature, To Know the Laws of Nature, Development of Technology Using the Laws	
Physical Quantities and		of Nature,	
Their Measurements	P-02	Physical Quantities and Their Measurements, Units of Measurements, Prefix, Dimension, Scientific Symbols and Notations Measuring Instruments, Scale or Ruler, Balance, Stop Watch, Vernier Calipers, Screw Gauge, Error and Accuracy.	
Chapter-03	P-11	Inertia and Concept of Force: Newton's First Law, Inertia, Force	
Force	P-12	Nature of Fundamental Force, Gravitational Force, Electromagnetic Force, Weak Nuclear Force or Weak force, Strong Nuclear Force, Balanced and Unbalanced Forces, Momentum	

<u>Chemistry</u>		
<u>Chapter</u>	<u>Lecture</u>	<u>Syllabus</u>
Chapter-02	C-03	Matter & States of matter, *Intermolecular force & energy, atomic mass (chart) + molecular mass.
States of Matter	C-04	Kinetic theory of particles & *postulates of kinetic theory, Diffusion, Effusion.

C-03     C-04     C-05     C-05 <t< th=""><th>applying heat &amp; mathematical explanation.     i   Heating &amp; cooling curve due to application of heat, Sublimation curve, Diffusion, Effusion (Revision).     i   Elements and Compounds, Atoms and Molecules, Symbols of Elements, Formula, The fundamental particles of an atom, Atomic Number, Mass Number.     i   Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     i   Bohr's Atomic Model, Success and Limitations of Bohr's Model.     i   Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     i   The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     i   Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotope.     i   Introduction to chemistry, the scopes of chemistry, Relationship between chemistry &amp; other branches of science, the importance of studying chemistry.     i   The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Wath</th></t<>	applying heat & mathematical explanation.     i   Heating & cooling curve due to application of heat, Sublimation curve, Diffusion, Effusion (Revision).     i   Elements and Compounds, Atoms and Molecules, Symbols of Elements, Formula, The fundamental particles of an atom, Atomic Number, Mass Number.     i   Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     i   Bohr's Atomic Model, Success and Limitations of Bohr's Model.     i   Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     i   The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     i   Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotope.     i   Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     i   The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Wath
Chapter-03     C-02       Chapter-03     C-02       Structure of Matter     C-12       Structure of Matter     C-12       Chapter-01     C-02       Chapter-01     C-02       Concepts of Chemistry     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Concepts of Chemistry     C-02       Chapter     M-02	Heating & cooling curve due to application of heat, Sublimation curve, Diffusion, Effusion (Revision).     Elements and Compounds, Atoms and Molecules, Symbols of Elements, Formula, The fundamental particles of an atom, Atomic Number, Mass Number.     Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     Bohr's Atomic Model, Success and Limitations of Bohr's Model.     Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals. <b>Wath</b>
Chapter-03     C-02       Chapter-03     C-10       Structure of Matter     C-10       Chapter-01     C-12       Chapter-01     C-02       Concepts of Chemistry     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Chapter     C-02       Concepts of Chemistry     C-02       Chapter     M-02	Elements and Compounds, Atoms and Molecules, Symbols of Elements, Formula, The Fundamental particles of an atom, Atomic Number, Mass Number.     A Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     Bohr's Atomic Model, Success and Limitations of Bohr's Model.     Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.
Chapter-03     C-02       Structure of Matter     C-10       Structure of Matter     C-10       Chapter-03     C-10       Chapter-01     C-10       Concepts of Chemistry     C-00       Concepts of Chemistry     C-00       Chapter     C-00       Concepts of Chemistry     C-00       Chapter     M-00	particles of an atom, Atomic Number, Mass Number.     Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     Bohr's Atomic Model, Success and Limitations of Bohr's Model.     Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals. <u>Math</u>
Chapter-03     C-05       Structure of Matter     C-10       Structure of Matter     C-11       Chapter-01     C-12       Chapter-01     C-07       Concepts of Chemistry     C-07       Chapter     C-07       Chapter     C-07       Chapter     C-07       Concepts of Chemistry     C-07       Chapter     M-07	Atomic Model, Rutherford's Atomic Model, Limitations of Rutherford's Model.     Bohr's Atomic Model, Success and Limitations of Bohr's Model.     Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Math
Chapter-03   C-10     Structure of Matter   C-11     Structure of Matter   C-11     Chapter-01   C-12     Concepts of Chemistry   C-02     Chapter O1   C-02     Concepts of Chemistry   C-02     Chapter O1   C-02     Chapter O1   C-02     Chapter O1   C-02	Bohr's Atomic Model, Success and Limitations of Bohr's Model.     Orbital Electronic Configuration of Atoms, Concept of Energy Sublevel, The Principles of Electronic     Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic     Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element     from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive     Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of     Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of     science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry     laboratory and in use of chemicals.
Chapter-03   Structure of Matter   C-11   C-12   Chapter-01   Concepts of Chemistry   C-02   Chapter   Chapter   Lectur   M-02	Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic     Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element     from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive     Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of     Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of     science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry     Iaboratory and in use of chemicals.     Math
Chapter-03 Chapter-03   Structure of Matter C-11   Chapter-01 C-07   Concepts of Chemistry C-07   Concepts of Chemistry C-07   Chapter C-07   Chapter C-07   Concepts of Chemistry C-07   Chapter C-07   Chapter C-07   Concepts of Chemistry C-07	Configuration in Atoms, Example.     The Principles of Electronic Configuration in Atoms (Revision), Some Exceptions in Electronic     Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element     from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive     Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of     Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of     science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry     Iaboratory and in use of chemicals.     Math
C-11 Chapter-01 Concepts of Chemistry Concepts of Chemistry Concepts of Chemistry C-02 Concepts of Chemistry C-02 Chapter	Configuration.     Isotopes, Atomic Mass or Relative Atomic Mass, Determining the Average Relative Mass of an Element     from Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive     Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of     Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches o     science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry     Iaboratory and in use of chemicals.
Chapter-01 C-02   Concepts of Chemistry C-02   Chapter Lecture   Chapter M-02	From Percentage of Isotope, Getting the Relative Molecular Mass from Relative Atomic Mass, Radioactive Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches o science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Image: Comparison of the scope of the sco
Chapter-01 C-02   Concepts of Chemistry C-02   Chapter Lecture   Chapter M-02	Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches o science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Math     Ce   Syllabus
Chapter-01 C-02   Concepts of Chemistry C-02   Chapter Lecture   Chapter M-02	Isotopes and Their Uses, Medical Science, Agriculture Sector, Generation of Electricity, Impact of     Radioactive Isotope.     Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Image: Comparison of the science of studying chemistry of the science of the sc
Chapter-01 Concepts of Chemistry C-02 Chapter Lectur M-02	Introduction to chemistry, the scopes of chemistry, Relationship between chemistry & other branches of science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Image: Math result     Syllabus
Chapter-01 Concepts of Chemistry C-02 Chapter Lectur M-02	science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Image: Math state     Image: Syllabus
Chapter-01 Concepts of Chemistry C-02 Chapter Lectur M-02	science, the importance of studying chemistry.     The process of research in chemistry, Steps in research in chemistry, Safety measures in chemistry laboratory and in use of chemicals.     Image: Math     Image: Comparison of the studying chemistry in the start of the star
C-02 Chapter Lectur M-02	Iaboratory and in use of chemicals.   Math   re Syllabus
M-03	<u>Math</u> re <u>Syllabus</u>
M-03	re <u>Syllabus</u>
M-03	
	Algebraic Expressions, Algebraic Formulae, Examples of 3.1, Exercises – 3.1 (1, 2).
M-1(	
	Exercis <mark>e 3.1 (3-15).</mark>
M-1	Formula <mark>of Cubes, Coroll</mark> ary, Works, Exercises - 3.2 (1, 2).
M-12	Exercise – 3.2 (3-15).
Chapter-03	Reducing Fractions, Fractions with Common Denominators, resolving into factors, Techniques for
M-13 Algebraic Expression	determining factors, work, Exercise-3.3 (1-15)
M-14	Exercise-3.3 (16-25).
	Exercise-3.3 (26-31), concept of remainder theorem, concept of factorization theorem, example, Work, Addition
M-1	and subtraction of algebraic fractions and equation.
M-10	6 Exercise-3.4 (1-16).
M-0	Classification of Real Numbers, Proof of Irrational Numbers, Decimal Fractions, Exercise-1 (9, 10, 20).
	Repeating Decimals, Conversion into Common Fractions, Addition and Subtraction of repeated Decimal
<u>Chapter-01</u> M-03	
Real Numbers	
M-0	
M-04	Square Roots, Infinite Decimal Fractions, Exercise-1 (11, 19, 21, 22).

<u>H. Math</u>		
<u>Chapter</u>	<u>Lecture</u>	<u>Syllabus</u>
<u>Chapter-02</u>	HM-09	Variables, Constant, Polynomials, Polynomials of One, Two and Three Variables, Cyclic, Symmetric and Homogenous Expressions, Activity on Page-40, Exercise-2 (1, 2), Example-22, Exercise-2 (d of 10 ).
Algebraic Expression	HM-10	Work (a) on Page-57, Work on Page-55, Exercise-2 (1,2,3 of 10).

	HM-11	Multiplication and Division of polynomial, Quotient and Product Theorem, Converse of factor theorem, Activity on Page-51-52, Exercise-2 (3-7), HW: Exercise-2 (15)		
	HM-12	Page-52 Proposition-1, Activity on Page-56 (2, 3), Example-18, Exercise-2 (8, 9, 12,13)		
	HM-13	Partial fractions, examples (23-29), Activity.		
	HM-14	Exercise-2 (11, 14).		
<u>Chapter-03</u>	HM-15	Projection of a point, Orthogonal projection, Theorem-1, 2, 3, 4		
	HM-16	Exercise – 3.1 (1, 2, 3, 4, 6).		
	HM-17	All Theorems of Apollonius, Theorem-5, Relationship between Side-Median, Exercise-3.1 (5, 7).		
Geometry	HM-18	Orthocenter, Circumcenter, Centroid, Nine Point Circle, Theorem- 6, 10, Exercise- 3.2 (8, 9), HW- 3.2 (16).		
	HM-19	Theorem- 7, 8, 9, 11, 12		
	HM-20	Exercise-3.2 (7, 10-14), HW- <mark>3.2 (15)</mark>		

<u>Biology</u>		
<u>Chapter</u>	<u>Lecture</u>	<u>Syllabus</u>
	B-03	Living cell, Types of cells, Plant cell and animal cell (With figure), Difference between plant cell and animal cell, Main organelles of plant and animal cells and their functions (Cell wall).
	B-04	Main organel <mark>les of plant</mark> and animal cells and their functions (Cell wall) (Protoplasm, Plasmalemma, Cytoplasmic organelles, Mitochondria).
Chapter-02	B-05	Plastid, Chloroplast, Chromoplast, Leucoplast, Golgi body, Endoplasmic reticulum, Cell vacuole, Lysosome.
Cells and Tissues of	B-06	Non-m <mark>embra</mark> nous cytoplasmic organelles, Nucleus.
Organisms	B-07	Roles of different cells in proper functioning of plants and animals, Plant tissue (Simple tissue, Complex tissue, Xylem, Phloem).
	B-08	Plant tissue (Epithelial tissue)
	B-09	Animal tissues (Connective tissue and its classification), Muscular tissue, Nerve tissue.
	B-10	Organ and system, Microscope, Electron microscope, Differentiations.
<u>Chapter-01</u>	B-01	Concept of biology, Branches of biology, Physical biology, applied biology, Classification of living beings, aim of classification, Living world (Margulis + R.H. Whittaker's classification- Monera, Protista).
Lessons of Life	B-02	Living world (Margulis + R.H. Whittaker's classification- Fungi, Plantae, Animalia), Different steps of classification, System of Binomial Nomenclature, Binomial names.
Chapter-03	B-11	Cell division and its classification, Mitosis, Stages of mitosis.
Cell Division	B-12	Significance of mitosis, Meiosis, Significance of meiosis, Difference between mitosis and meiosis, Discussion about haploid and diploid cells.

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